LETTERS TO THE EDITOR

Phosphorylation of Optically Pure Salicylal-1,2-diphenylethanediamine with Alkylene Chlorophosphites

L. K. Kibardina, A. R. Burilov, and M. A. Pudovik

Arbuzov Institute of Organic and Physical Chemistry, Kazan Scientific Center, Russian Akademy of Sciences, ul. Arbuzova 8, Kazan, 420088 Russia

e-mail: pudovik@iopc.knc.ru

Received November 1, 2011

DOI: 10.1134/S1070363212030279

The reaction of salicylalethylenediimine with 2-chloro-1,3,2-dioxaphospholane was found to be a cascade process, resulting in the framework derivatives of hexacoordinated phosphorus atom involving the intramolecular transannular N→P bond (by X-ray data) [1]. It is important that the reaction is highly stereoselective and gives rise to a single diastereomer. Aiming to obtain the optically pure derivative of hexacoordinated phosphorus atom, diimine I {obtained in

the reaction of 1S,2S(-)-diphenylethanediimine with salicylaldehyde [2]} was introduced into the reaction with ethylene chlorophosphite. Diimine I reacts with ethylene chlorophosphite IIa to give an optically pure diastereomer IIIa in 82% yield. Its phosphorus chemical shift is δ_P –113.32 ppm, which is characteristic of P(VI) derivatives. The optical rotation of compound IIIa is $[\alpha]_D^{20}$ –28.1° (c 0.3739, CH₂Cl₂).

Under the treatment with triethylamine, compound **IIIa** undergoes dehydrochlorination to form phosphorate **IVa** of a neutral structure $\{\delta_P - 101.52 \text{ ppm}, [\alpha]_D^{20} - 29.4^{\circ} (c \ 0.3989, CH_2Cl_2)\}.$

The reaction with 1,2-propylene chlorophosphite **IIb** leads to the appearance of another chiral center and to the formation of diastereomers mixture **IIIb** (δ_P –112.34, 112.99 ppm) in 1:1.4 ratio.

1,1-Ethylenedioxy-3,4,11,12-dibenzo-6-aza-9-ammonium-7,8-diphenyl-2,13-dioxa-1-phospha-[8.3.0^{1,10}]tridecatri-3,5,11-ene chloride (IIIa). A mixture of 0.42 g of diimine I and 0.13 g of chloride IIa in 10 ml of methylene chloride was kept for 7 days at 20°C. The solvent was removed and the residue was washed with diethyl ether. Yield 0.45 g (82%), mp 164–167°C, $[\alpha]_D^{20}$ –28° (c 0.3739, CH₂Cl₂). IR spectrum, v, cm⁻¹: 1624 (C=N), 2415–2678 (NH₂+). ³¹P NMR spectrum

(CDCl₃): δ_P –113.32 ppm. Mass spectrum (MALDITOF), m/z: 510 [M^+ – HCl]. Found, %: N 5.31; P 6.02. $C_{30}H_{28}ClN_2O_4P$. Found, %: N 5.12; P 5.67.

1,1-(1,2-Propylenedioxy)-3,4,11,12-dibenzo-6-aza-9-ammonium-7,8-diphenyl-2,13-dioxa-1-phospha-[8.3.0^{1,10}]tridecatri-3,5,11-ene chloride (IIIb) was similarly obtained from 0.42 g of diimine I and 0.14 g of chloride IIa. Yield 0.46 g (89%), mp 149–151°C. IR spectrum, ν , cm⁻¹: 1628 (C=N), 2413–2669 (NH₂⁺). ³¹P NMR spectrum (CDCl₃), δ_P , ppm: -112.34, -112.99. Mass spectrum (MALDI-TOF), m/z: 524 [M^+ – HCl]. Found, %: N 5.01; P 5.34. C₃₁H₃₀ClN₂O₄P. Calculated, %: N 4.99; P 5.52.

1,1-Ethylenedioxy-3,4,11,12-dibenzo-6,9-diaza-7,8-diphenyl-2,13-dioxa-1-phospha[8.3.0^{1,10}]tridecatri-3,5,11-ene (IVa). A mixture of 0.14 g of compound IIIa and 0.06 g of triethylamine in 10 ml of methylene chloride was refluxed for 2 h. After the removal of the solvent the reaction mixture was diluted with benzene, filtered off from triethylamine hydrochloride, and concentrated. Yield 0.26 g (71%), mp 146°C. IR spectrum, v, cm⁻¹: 1635 (C=N), 3317 (NH). ³¹P NMR spectrum (CDCl₃): δ_P –101.52 ppm. Mass spectrum

(MALDI-TOF), *m/z*: 510. Found, %: N 5.22; P 6.13. C₃₀H₂₇N₂O₄P. Calculated, %: N 5.49; P 6.07.

The IR spectra were recorded on a Bruker Vector 22 Fourier-spectrometer from KBr pellets in a range of 400–4000 cm⁻¹. The ³¹P NMR spectra were registered on a Bruker MSL-400 Fourier-spectrometer (100.62 MHz). The mass spectra (MALDI-TOF) were taken on a Ultraflex III TOF/TOF Bruker instrument using *p*-nitroaniline as a matrix.

ACKNOWLEDGMENTS

This work was financially supported by Russian Foundation for Basic Research (grant no. 12-03-00204).

REFERENCES

- 1. Kibardina, L.K., Terent'yeva, S.A., Kataeva, O.N., Burilov, A.R., and Pudovik, M.A., *Zh. Obshch. Khim.*, 2010, vol. 80, no. 2, p. 341.
- Yaozhong Jiang, Liuzhu Gong, Xiaoming Feng, Wenhao Hu, Weidong Pan, Zhi, L., and Aiqiao, M., Tetrahedron, 1997, vol. 53, no. 42, p. 14327.